TED (10)-1016B

(REVISION-2010)

Reg. No.

Signature

SECOND SEMESTER DIPLOMA EXAMINATION IN ENGINEERING/ TECHNOLOGY-MARCH, 2012

APPLIED SCIENCE-II (CHEMISTRY)

(Common except DCP and CABM)

[Time : 11/2 hours

(Maximum marks : 50)

Marks

2

2

[P.T.O.

PART-A

(Answer the following questions in one or two sentences. Each question carries 2 marks.)

- I Give reason and explain the following observations :
 - Aqueous solution of raw sugar becomes colourless when passed over animal charcoal.
 - 2. Air becomes dry in presence of silica gel.

PART-B

(Answer a	iny two f	ull questions.	Each question	carries 8 marks.)	

П	1.	Reduction potentials of Zn and Ag electrodes are -0.76 V and 0.80 V respectively:		
		(a) Which one is a better oxidizing agent ?		
		(b) Which metal displaces the other metal from its solution?	4	
	2.	(a) What is the effect of temperature on electric conduction on metallic and electrolytic conductors?		
		(b) Rate of corrosion is more near to seashore area. Explain your reason.	4	
ш	I.	How will you represent Daniel cell? Write the electrode reactions and cell reaction.		
	2.	What is London smog? What are the environmental impact of photochemical smog?		
IV	1.	Identify the functional group in the following molecules :		
		(a) CH ₃ COOCH ₃ ; CH ₃ O CH ₃		
		(b) Write the possible isomers of C_4H_{10} .	4	
	2.	Illustrate with two examples :		
		(a) Saturated and unsaturated hydrocarbons.		
		(b) Condensation polymerization.	4	

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PART-C

(Answer one full question from each unit. Each question carries 15 marks.)

UNIT-I

V	1.	Physisorption can be changed to chemisorption. Illustrate with examples.	4
	2.	(a) Explain the electrochemical theory of corrosion.	
		(b) Galvanization.	4
	3,	Write down the cell reaction Ni/Cd cell. How do represent it?	4
	4.	Sketch O2-H2 fuel cell and cell reaction involved in it.	3
		OR	
V1	1.	Explain the electrolysis of molten NaCl.	4
	2.	How are electrochemical cell classified and compare them with examples ?	4
	3.	What are different types of barrier protections ?	4
	4.	Which part of the iron corrodes if it partially immersed in an electrolyte? Give reason.	3

UNIT-II

VII	1.	(a) How you distinguish between saturated and unsaturated hydrocarbon ?	
		(b) Give two examples of nuclear fuels.	4
	2.	(a) Mention the process used to improve the properties of rubber.	
		(b) What are the monomers in Buna-S and Buna-N ?	4
	3.	 High boiling gasoline fractions are not good fuels. Suggest a suitable process to convert them into good fuels. 	
	4.	How does ozone layer depletion happens?	3
		Or	
VIII	1.	What are different type of polymerizations ? Give examples.	4
	2.	What are the differences between thermoplastic and thermosetting plastic ?	4
	3.	What is mean by propellants? How are they classified?	4
	4.	How does rain becomes acid rain?	3